

## High Pressure Synthesis and Electrical and Magnetic Properties of $\text{MnGe}_4$ and $\text{CoGe}_4$

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The new germanides,  $\text{MnGe}_4$  and  $\text{CoGe}_4$ , were synthesized by solid state reactions at 5.5–6 GPa and 600–700°C for 2 hr using the belt-type high pressure apparatus. The crystal structures of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  are superstructures of the  $\beta\text{-NiHg}_4$ -type structures. The unit cell of  $\text{MnGe}_4$  is composed of four  $\beta\text{-NiHg}_4$ -type cells and that of  $\text{CoGe}_4$  is composed of eight  $\beta\text{-NiHg}_4$ -type cells.  $\text{MnGe}_4$  is a metallic conductor and an itinerant electron ferromagnet with a Curie temperature of 340 K.  $\text{CoGe}_4$  is a metallic conductor and a Pauli paramagnet. © 1990 Academic Press, Inc.

### Introduction

There have been a number of investigations about the preparations and the physical properties of intermetallic compounds in the binary transition metal–4B metalloid atom systems. Transition metal silicides, especially, have been widely investigated because of their various kinds of electrical and magnetic properties and their potential application in solid state electronic devices (1, 2).

Although crystal structure and physical properties of transition metal germanides are expected to be similar to those of transition metal silicides, no systematic investigations have yet been carried out. Transition metal germanides have various chemical compositions represented by formulas such as  $T_3\text{Ge}$ ,  $T_2\text{Ge}$ ,  $T_5\text{Ge}_3$ ,  $T_{11}\text{Ge}_8$ ,  $T\text{Ge}$ ,  $T\text{Ge}_2$ , and  $T\text{Ge}_4$  ( $T$ : transition metal). Among these germanides, the compounds containing more than 67 at.% Ge were rarely investigated. Only two compounds,  $\text{IrGe}_4$  and

$\text{RhGe}_4$ , were synthesized (3, 4). Recent investigations on synthesis of transition metal germanides demonstrate that the high pressure synthesis is a suitable method for the preparation of new germanides with high germanium contents (5–7).

In the present study,  $\text{MnGe}_4$  and  $\text{CoGe}_4$  were synthesized under high pressure–temperature conditions, and their electrical and magnetic properties were investigated.

### Experimental

Manganese and cobalt powders (>99.9% purity) and germanium powder (>99.99% purity) were mixed with Ge/ $T$  ( $T$ : Mn, Co) atomic ratios varying from 4.0 to 5.0 using an agate mortar and uniaxially pressed at 100 MPa at room temperature to form pellets, 5 mm in diameter and 3 mm in thickness. The pellets were put into a cylindrical BN capsule, which was placed in a carbon heater. The assemblage was put into a cell constructed with NaCl and subjected to high

pressure-temperature conditions using the belt-type high pressure apparatus. The high pressure-temperature treatments were carried out at 5.5–6 GPa and 600–700°C for 2 hr, and then the samples were quenched to room temperature prior to releasing the applied pressure. The detailed preparation procedures were described in the previous paper (5).

The pellets thus obtained were pulverized and the residual germanium was removed by leaching with 5 N NaOH + 3% H<sub>2</sub>O<sub>2</sub> solutions at room temperature. Chemical analysis was carried out by means of ICP emission spectrochemical analysis.

The phase of product was identified by X-ray powder diffraction analysis using Ni-filtered CuK $\alpha$  radiation or Fe-filtered CoK $\alpha$  radiation. Lattice parameters were determined by a least-squares method using silicon as an internal standard.

The electrical resistivity was measured by a standard four-probes method in the temperature range 77–400 K. Magnetization and magnetic susceptibility were measured by a magnetic torsion balance in magnetic field up to 10 kOe in the temperature range 77–450 K.

**Experimental Results**

*MnGe<sub>4</sub>*

MnGe<sub>4</sub> was synthesized at 5.5 GPa and 600–700°C for 2 hr. The product obtained from the reaction with the starting composition of Ge/Mn = 4.0 contained small amounts of Mn<sub>3</sub>Ge<sub>5</sub> and MnGe<sub>2</sub>. A single phase of MnGe<sub>4</sub> was obtained from the product with the starting composition of Ge/Mn = 4.8 by leaching out the residual germanium. The germanium content of the single phase sample was analyzed to be 85.8 ± 0.1 wt% Ge, which was relatively higher than the calculated value of 84.1 wt% Ge in MnGe<sub>4</sub>.

The X-ray powder diffraction data of

TABLE I  
X-RAY POWDER DIFFRACTION DATA OF MnGe<sub>4</sub>

<i>d</i> <sub>obsd</sub> (nm)	<i>d</i> <sub>calcd</sub> (nm)	<i>h</i>	<i>k</i>	<i>l</i>	<i>I</i> <sub>obsd</sub>
0.551	0.5515	2	0	0	vw
0.500	0.4992	1	0	1	vw
0.3072	0.3073	3	0	1	m
0.2800	0.2799	0	0	2	s
0.2757	0.2757	4	0	0	vs
0.2496	0.2496	2	0	2	w
0.2469	0.2466	4	2	0	w
0.2416	0.2414	4	1	1	w
0.2051	0.2052	4	3	1, 5 0 1	vw
0.1965	0.1964	4	0	2	vvs
0.1951	0.1950	4	4	0	s
0.1839	0.1839	1	0	3, 6 0 0	w
0.1600	0.1600	4	4	2	m
0.1425	0.1425	4	3	3, 5 0 3	w
0.1400	0.1400	0	0	4	w
0.1378	0.1378	5	2	3, 8 0 0	w
0.1342	0.1342	6	4	2	w
0.1329	0.1329	7	4	1, 8 1 1	vw
0.1265	0.1266	5	4	3	vw
0.1248	0.1248	4	0	4	w
0.1237	0.1237	8	0	2	w
0.1234	0.1233	6	3	3, 8 4 0	w
0.1204	0.1204	7	0	3	vw
0.1138	0.1137	4	4	4	w
0.1129	0.1128	8	4	2	m
0.1113	0.1114	1	0	5, 6 0 4	vw
0.1104	0.1103	7	4	3, 8 1 3, 10 0 0	w
0.1072	0.1071	3	0	5	vw
0.1033	0.1033	4	1	5, 6 4 4	vw
0.1026	0.1026	8	6	2, 10 0 2	vw
0.1024	0.1024	9	0	3, 10 4 0	vw

MnGe<sub>4</sub> are listed in Table I. All diffraction peaks of MnGe<sub>4</sub> could be completely indexed in the tetragonal structure with lattice constants of *a* = 1.103 ± 0.001 and *c* = 0.5598 ± 0.0003 nm, respectively. Since the observed reflections satisfied the condition of *h* + *k* + *l* = 2*n*, the space group is expected to be one of *I4/m*, *I422*, *I4mm*, *I4m2*, *I42m*, or *I4/mmm*.

MnGe<sub>4</sub> was metastable under ambient pressure condition and decomposed into Mn<sub>3</sub>Ge<sub>5</sub> and Ge at 270–300°C.

The results of magnetization measurements of MnGe<sub>4</sub> are shown in Figs. 1 and

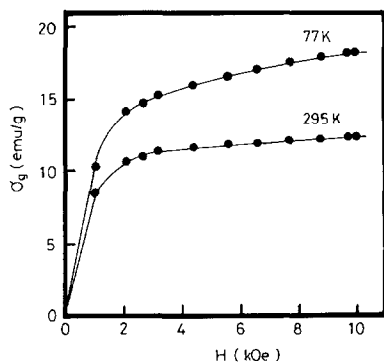


FIG. 1. Magnetic field dependence of magnetization of  $\text{MnGe}_4$ .

2.  $\text{MnGe}_4$  exhibited ferromagnetic behavior with a Curie temperature of 340 K. The saturation magnetic moment,  $\mu_s$ , of Mn atom extrapolated to 0 K was calculated to be  $1.2 \pm 0.1 \mu_B$ . Above the Curie temperature, the magnetic susceptibility followed the Curie-Weiss-type law  $M\chi_g = M\chi_c + C/(T - \Theta_p)$ , where  $M$  is the formula weight and  $\chi_c$  is the temperature-independent magnetic susceptibility. The number of effective Bohr magnetons,  $P_{\text{eff}}$ , and the paramagnetic Curie temperature,  $\Theta_p$ , were calculated to be  $2.83 \pm 0.03 \mu_B$  and  $349 \pm 3$  K, respectively.

Figure 3 shows the results of the electrical resistivity measurement;  $\text{MnGe}_4$  exhibits metallic conduction. The  $\rho$ - $T$  curve was not

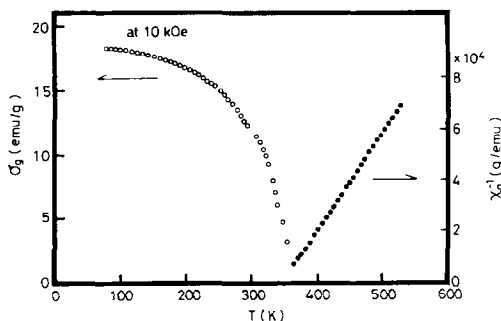


FIG. 2. Temperature dependence of magnetization and reciprocal magnetic susceptibility of  $\text{MnGe}_4$ .

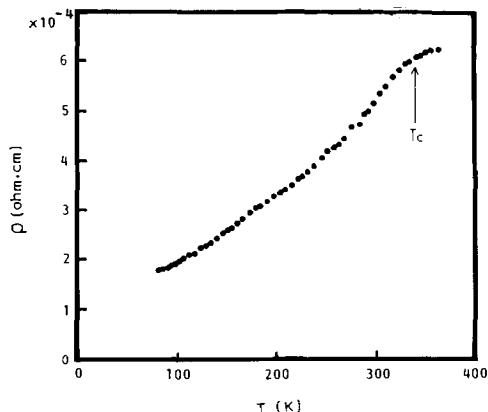


FIG. 3. Temperature dependence of electrical resistivity of  $\text{MnGe}_4$ .

linear in the measured temperature range; an anomaly was observed at the Curie temperature.

#### $\text{CoGe}_4$

$\text{CoGe}_4$  was synthesized at 6 GPa and 600–700°C for 2 hr. Single phase  $\text{CoGe}_4$  was obtained from the product with the starting composition of Ge/Co = 4.2 by leaching out the residual germanium. The germanium content of the single phase sample was analyzed to be  $83.5 \pm 0.1$  wt% Ge, which was slightly higher than the calculated value of 83.1 wt% Ge in  $\text{CoGe}_4$ .

The X-ray powder diffraction data of  $\text{CoGe}_4$  are listed in Table II. All diffraction peaks of  $\text{CoGe}_4$  were completely indexed in the cubic structure with a lattice constant of  $a = 1.099 \pm 0.001$  nm. Since the observed reflections satisfied the condition  $h + k + l = 2n$ , the space group is expected to be one of  $I23$ ,  $I2_13$ ,  $Im3$ ,  $I432$ ,  $I43m$ ,  $Ia3$ , or  $Im3m$ .

$\text{CoGe}_4$  was metastable under ambient pressure condition and decomposed into  $\text{CoGe}_2$  and Ge at 380–400°C.

The temperature dependence of electrical resistivity of  $\text{CoGe}_4$  is shown in Fig. 4.  $\text{CoGe}_4$  shows metallic conduction with a linear temperature dependence of resistivity.

TABLE II  
X-RAY POWDER DIFFRACTION DATA OF CoGe<sub>4</sub>

$d_{\text{obsd}}$ (nm)	$d_{\text{calcd}}$ (nm)	$h$	$k$	$l$	$I_{\text{obsd}}$
0.449	0.4487	2	1	1	vw
0.318	0.3172	2	2	2	vw
0.2938	0.2937	3	2	1	m
0.2749	0.2748	4	0	0	vs
0.2592	0.2590	3	3	0, 4 1 1	vw
0.2459	0.2457	4	2	0	m
0.2157	0.2155	4	3	1, 5 1 0	vw
0.2007	0.2006	5	2	1	w
0.1944	0.1943	4	4	0	vvs
0.1832	0.1832	6	0	0, 4 4 2	w
0.1784	0.1783	6	1	1, 5 3 2	w
0.1587	0.1586	4	4	4	m
0.1497	0.1496	5	5	2, 6 3 3, 7 2 1	vw
0.1374	0.1374	8	0	0	m
0.1334	0.1333	6	4	4, 8 2 0	vw
0.1229	0.1229	8	4	0	m
0.1122	0.1122	8	4	4	m

The measurements of the magnetic susceptibility show that CoGe<sub>4</sub> is a Pauli paramagnetic with a magnetic susceptibility of  $5 \times 10^{-7}$  emu/g.

**Discussion**

*Crystal Structures of MnGe<sub>4</sub> and CoGe<sub>4</sub>*

Some TX<sub>4</sub> compounds of the transition metal-4B metalloid atom systems have

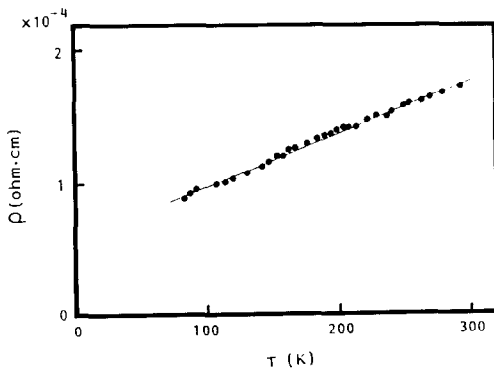


FIG. 4. Temperature dependence of electrical resistivity of CoGe<sub>4</sub>.

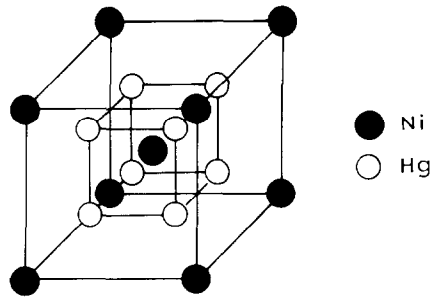


FIG. 5. Crystal structure of  $\beta$ -NiHg<sub>4</sub>.

been synthesized. From the crystallographic point of view, these compounds are classified into three groups. IrGe<sub>4</sub> has the hexagonal structure with the space group of P3<sub>1</sub>21, which is closely related to defect-type  $\gamma$ -brass structure (3). RhGe<sub>4</sub> is isostructural with IrGe<sub>4</sub> (4). Stannides (IrSn<sub>4</sub>, PdSn<sub>4</sub>, and PtSn<sub>4</sub>) have the orthorhombic PtSn<sub>4</sub>-type structure (space group Aba2), which can be derived from the CoGe<sub>2</sub>-type structure (4, 8-10). PtPb<sub>4</sub> has the tetragonal structure with the space group of P4/nbm, which is derived from the CuAl<sub>2</sub>-type structure (11). No further TX<sub>4</sub> compounds with other crystal structures than the above three types have been reported in the literature.

The crystal structures of MnGe<sub>4</sub> and CoGe<sub>4</sub> do not belong to the above-mentioned structure groups. The unit cells of MnGe<sub>4</sub> and CoGe<sub>4</sub> are considered to be superlattices of the  $\beta$ -NiHg<sub>4</sub>-type cell. Figure 5 shows the crystal structure of  $\beta$ -NiHg<sub>4</sub>. The  $\beta$ -NiHg<sub>4</sub> type is a filled up derivative of the bcc structure. Ni atoms form a bcc lattice and are surrounded by eight Hg atoms at the corners of a cube. The  $\beta$ -NiHg<sub>4</sub>-type structure is found in NiHg<sub>4</sub>, PtHg<sub>4</sub>, CrGa<sub>4</sub>, and MnGa<sub>4</sub> (12). Among these compounds, the atomic sizes of the constituent elements in MnGa<sub>4</sub> are almost the same as those in MnGe<sub>4</sub> and CoGe<sub>4</sub>. The comparison of the sizes of the unit cells of MnGa<sub>4</sub>, MnGe<sub>4</sub>, and CoGe<sub>4</sub> is shown in Fig. 6. The size of the unit cells of MnGe<sub>4</sub> and CoGe<sub>4</sub> corre-

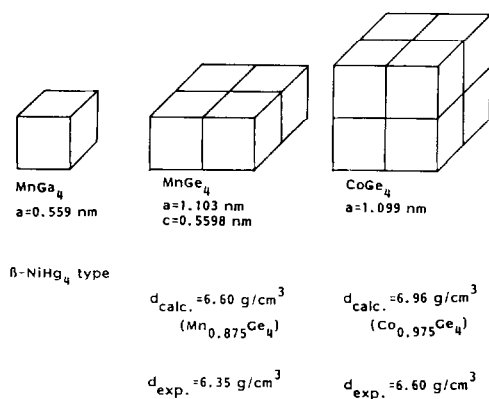


FIG. 6. Unit cells of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  compared with the  $\text{MnGa}_4$  structure ( $\beta\text{-NiHg}_4$  type).

spond to four times and eight times that of  $\text{MnGa}_4$ , respectively.

Since the  $\beta\text{-NiHg}_4$ -type cell contains 2 formula units, the unit cells of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  should contain 8 and 16 formula units, respectively. According to the present model, the theoretical densities of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  were calculated to be 6.60 and 6.96  $\text{g/cm}^3$ , respectively. The chemical formula units of the compounds were taken as  $\text{Mn}_{0.875}\text{Ge}_4$  and  $\text{Co}_{0.975}\text{Ge}_4$ , which follows the results of chemical analysis. The pycnometrically measured densities of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  were 6.35 and 6.60  $\text{g/cm}^3$ , which agree with the calculated values within 5%.

Figure 7 shows the X-ray powder diffraction patterns of  $\text{MnGa}_4$ ,  $\text{MnGe}_4$ , and  $\text{CoGe}_4$ . In the diffraction patterns of  $\text{MnGe}_4$  and  $\text{CoGe}_4$ , the diffraction peaks corresponding to the strong diffraction peaks of  $\text{MnGa}_4$  were also observed in  $\text{MnGe}_4$  with the  $\beta\text{-NiHg}_4$ -type structure. This fact suggests that the crystal structures of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  are the superstructures of the  $\beta\text{-NiHg}_4$ -type structure.

The absence of reflections (220) in the diffraction patterns of  $\text{MnGe}_4$  and  $\text{CoGe}_4$ , which correspond to the (110) peak of  $\text{MnGa}_4$ , indicate that the slight positional

changes of transition metal atoms from the ideal positions in the  $\beta\text{-NiHg}_4$ -type structure occur. It is plausible that the formation of the superstructures in  $\text{MnGe}_4$  and  $\text{CoGe}_4$  is due to vacancies at the transition metal sites and thus results in small positional changes of transition metal atoms. Since attempts of the single crystal growth of  $\text{MnGe}_4$  and  $\text{CoGe}_4$  under high pressure conditions were not successful, detailed structural analyses on single crystals were not possible.

#### Magnetic Properties of $\text{MnGe}_4$

$\text{MnGe}_4$  is a ferromagnetic with a Curie temperature of 340 K. The value of saturation magnetic moment,  $\mu_s$ , does not agree with that of paramagnetic moment,  $\mu_c$ , determined from the Curie-Weiss law. The curve of the magnetization against temperature is close to a Brillouin function and shows no evidence of ferrimagnetic behavior. The nonsaturating behavior of magnetization (see Fig. 1) can be explained by the itinerant electron theory of ferromagnetism (13, 14). In that case, the disagreement between  $\mu_s$  and  $\mu_c$  may be due to the spin fluctuation in the itinerant electron system.

In ferromagnets with the localized magnetic moment systems,  $\mu_c/\mu_s$  (Rhodes-Wohlfarth ratio) is 1. However, in an itinerant electron ferromagnet,  $\mu_c/\mu_s$  is larger than 1 (15). The relationship between  $\mu_c/\mu_s$  and  $T_c$  is characteristic of itinerant electron ferromagnets. The  $\mu_c/\mu_s$  values fall on a unique curve if plotted against the Curie temperature (15). Itinerant electron ferromagnets have wide distribution of properties, varying from "weakly ferromagnets," in which the  $T_c$  is low and the  $\mu_c/\mu_s$  is large, to "ferromagnets with nearly localized moment," in which the  $T_c$  is high and the  $\mu_c/\mu_s$  is nearly equal to 1 (15, 16).  $\text{MnGe}_4$  ( $\mu_c/\mu_s = 1.7$ ,  $T_c = 340 \text{ K}$ ) lies on the Rhodes-Wohlfarth curve and is situated in the intermediate region between weakly ferromagnets and nearly localized ferromagnets.

In itinerant electron systems, spin fluctu-

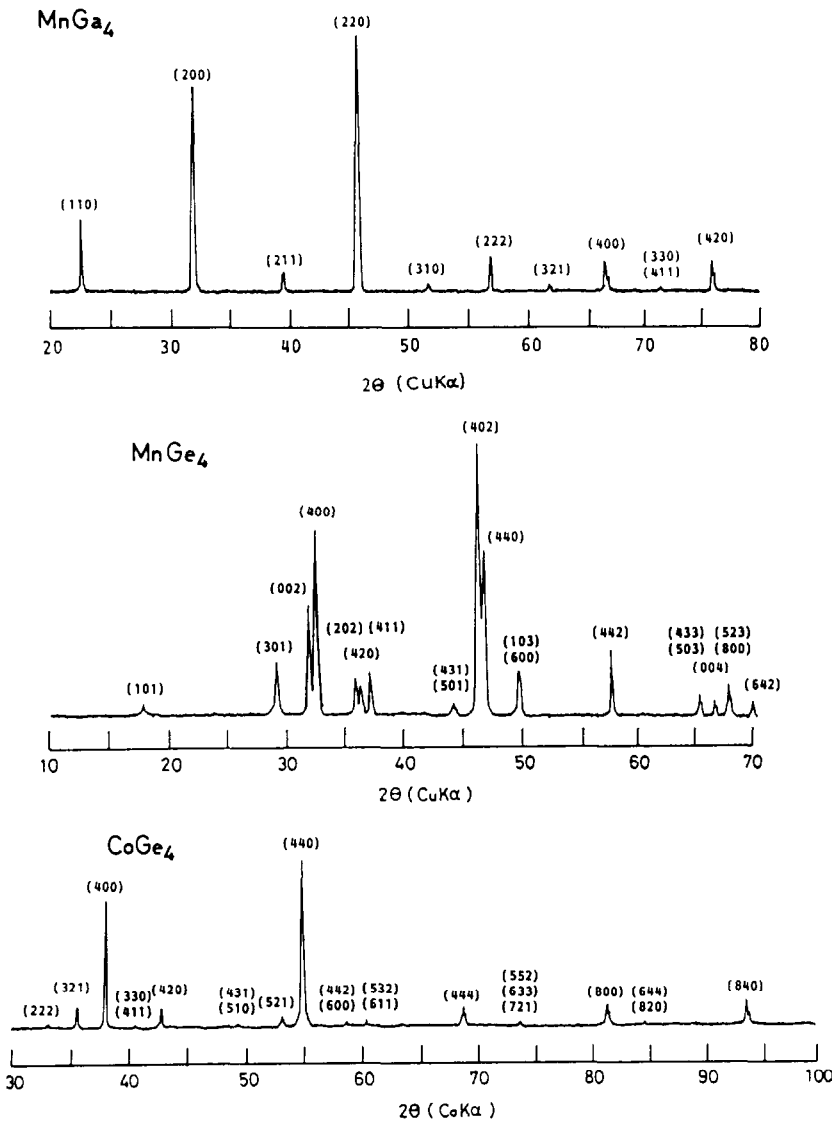


FIG. 7. X-ray powder diffraction patterns of  $\text{MnGa}_4$ ,  $\text{MnGe}_4$ , and  $\text{CoGe}_4$  ( $\text{CuK}\alpha$  radiation for  $\text{MnGa}_4$  and  $\text{MnGe}_4$ ,  $\text{CoK}\alpha$  for  $\text{CoGe}_4$ ).

ations have an important effect upon the magnetic and transport properties of the compounds (17–19). A theory for the transport properties of weakly ferromagnets has been developed (18). However, no satisfactory theory has been proposed for itinerant electron ferromagnets belonging to the in-

termediate region between weak ferromagnets and nearly localized ferromagnets. The electrical resistivity,  $\rho$ , of a weak ferromagnet displays the relation  $\rho$  vs  $T^2$  at low temperature and  $\rho$  vs  $T^{5/3}$  at near  $T_c$  due to spin fluctuations (18). Although  $\text{MnGe}_4$  is not really a weakly ferromagnet, a  $T^{5/3}$  depen-

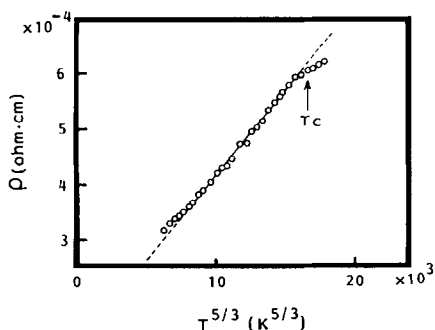


FIG. 8.  $\rho$  versus  $T^{5/3}$  plots of  $\text{MnGe}_4$ .

dence of the electrical resistivity was observed in the temperature region near  $T_c$  as shown in Fig. 8. This fact might be an indication that the effect of spin fluctuations on the transport predominates, relative to the lattice vibration effects in  $\text{MnGe}_4$ .

### Summary

Two new germanides,  $\text{MnGe}_4$  and  $\text{CoGe}_4$ , were synthesized under high pressure-temperature conditions. The exact chemical compositions of the prepared specimens were determined to be  $\text{Mn}_{0.875}\text{Ge}_4$  and  $\text{Co}_{0.975}\text{Ge}_4$ , respectively. The crystal structures of both compounds were superstructures of the  $\beta$ - $\text{NiHg}_4$ -type structure. The unit cell of  $\text{MnGe}_4$  is composed of four  $\beta$ - $\text{NiHg}_4$ -type cells, and that of  $\text{CoGe}_4$  is composed of eight  $\beta$ - $\text{NiHg}_4$ -type cells. The formation of the superstructure might originate from the defect formation at the transition metal sites and from the slight positional changes of the transition metal atoms. Iron ferromagnet with a Curie temperature of 340 K, and is situated in the intermediate region between weakly ferromagnets and

nearly localized ferromagnets on the Rhodes-Wohlfarth curve.  $\text{CoGe}_4$  is a Pauli paramagnetic metal.

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